



Chemical Analysis of metal samples using Optical Emission Spectroscopy (OES)

Glen Thiele Spectrosource Pty Ltd



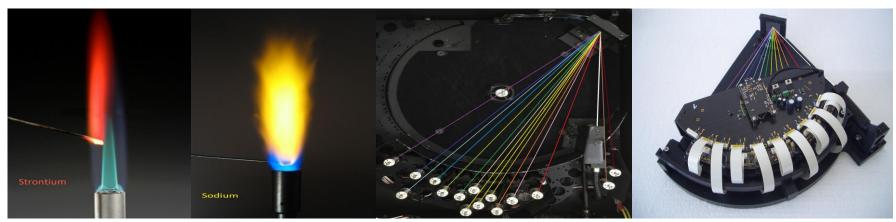








What is Optical emission spectroscopy (OES) also known as Atomic emission spectroscopy (AES)?

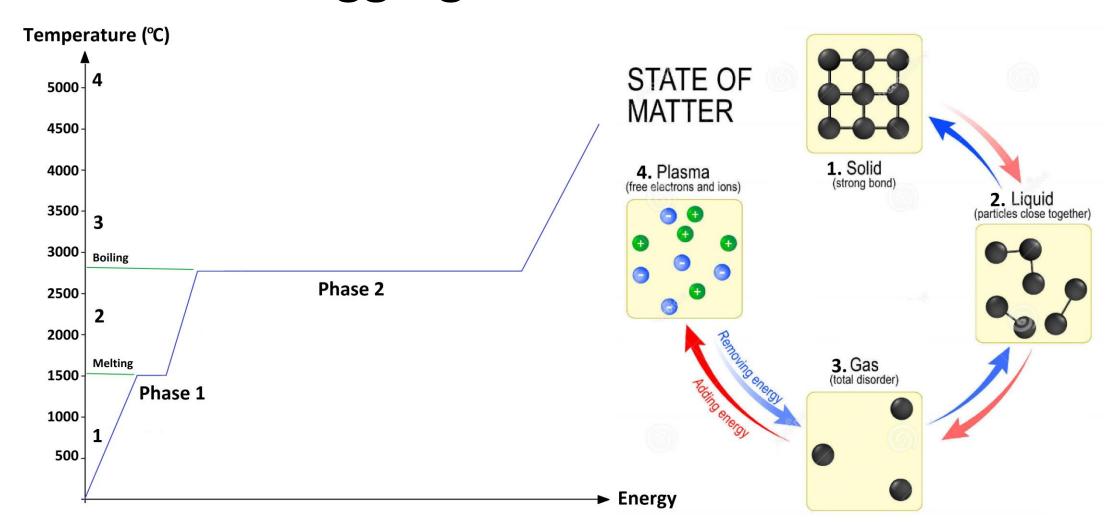






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Aggregate state of matter







Atomic physics – How photons are emitted

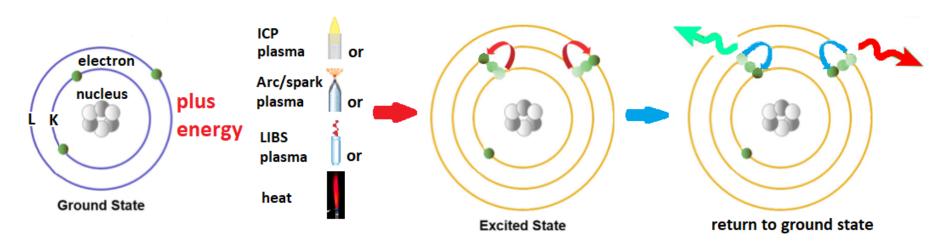
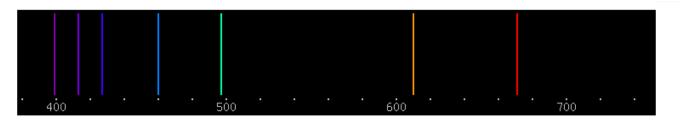
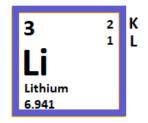


Figure 1: Bohr model (1913) Lithium --> excitation --> relaxation --> photon emission





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Figure 2: Lithium emission spectrum





Optical emission Spectrometers







ICP OES

LIBS OES

Arc/Spark OES





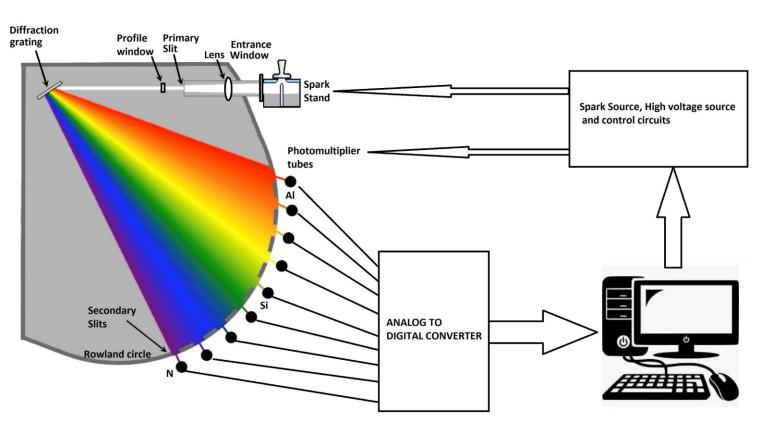
Why Use Spark OES Analysis?

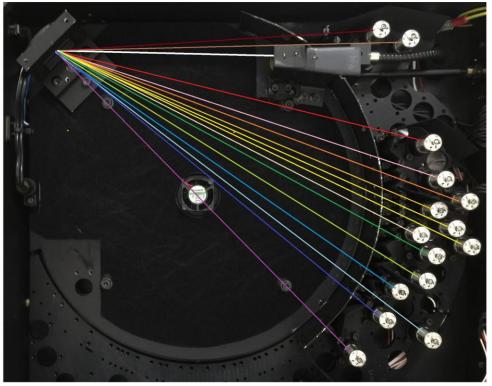
- Most widely used, reliable and reproducible metal analysis technique.
- Wide choice of instruments.
- Excellent LOD
- CCD based spectrometers allow a high number of elements and bases.
- Detection of N is possible with high end portable and high end laboratory spectrometers.
- Detection of O is possible with high end laboratory spectrometers.





PMT based Spark OES architecture

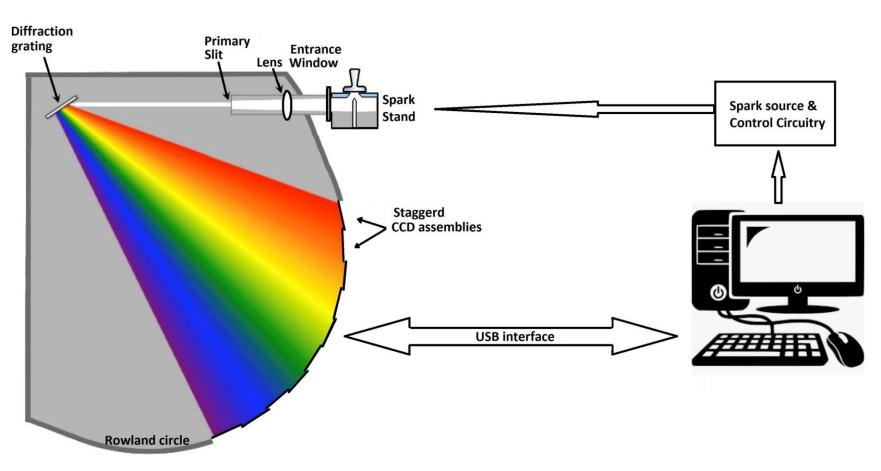


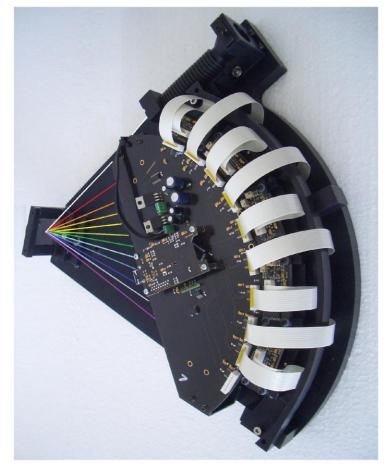






CCD based Spark OES architecture

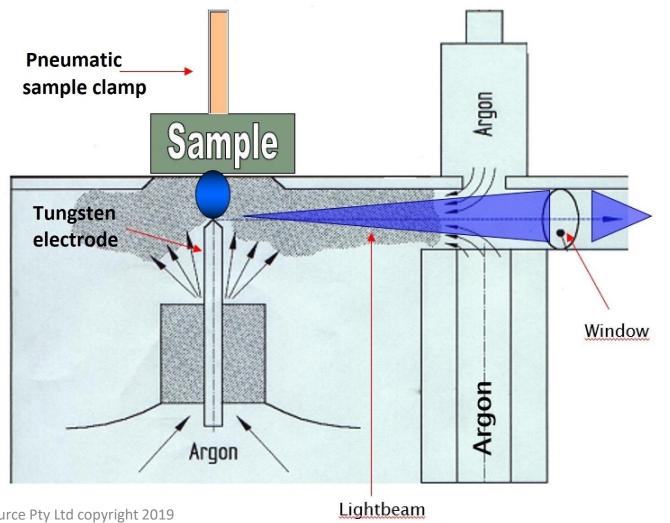








Spark Stand architecture

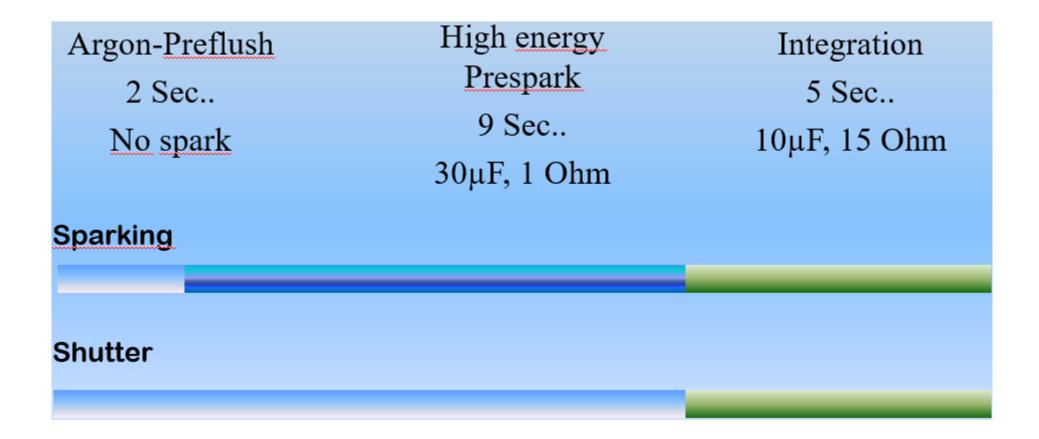








Typical spark sequence



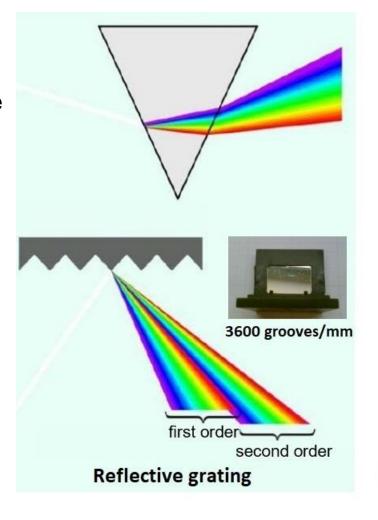


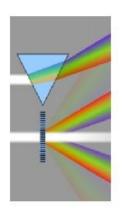


Grating properties

Prism:

- non-linear refraction
- affected by temperature due to influence of medium density







Transmission grating

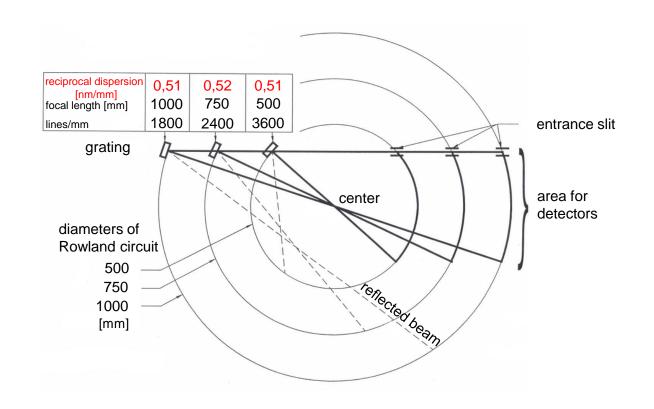
Grating:

- linear diffraction
- different orders





Reciprocal dispersion as criteria of line separation



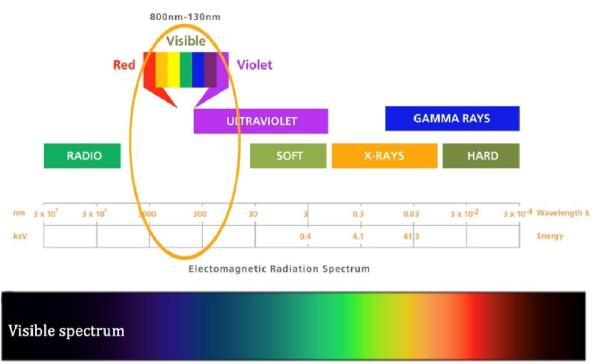


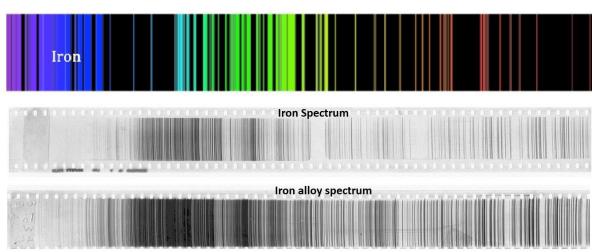
3600 grooves/mm grating





Spark OES Spectrum 130nm – 800nm



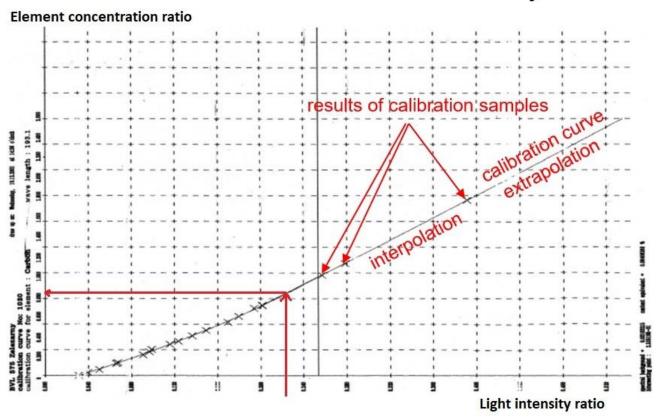






Calculation of results from Calibration Curve

Calibration curve for Carbon in low alloy steel



16 MnCr ! 0.30 1.26 0,015 0.022 0.09 0.06 0.18 0.106 0.18 0.30 1.26 0.015 0.022 0.09 0.106 1.06 0.06 0.18 1,26 0.015 0,022 0.09 0.06 0.30 0.106 0.18 0.30 1.26 0.015 0.022 0.09 1.16 0.106 0.06 0.18 1.26 0.015 0.022 0.09 1.09 0.06 0.30 0.106 0.18 0.30 0.015 0.022 0.106 0.06 0.06 0.18 0.30 1.26 0,015 0.022 0.09 0.106 1.08 0.04 0.02 0.029 0.003 0.039 0.05 0.06 0.04 0.02 0.039 0.05 0.06 0.029 0.003 0.039 0.05 0.06 0.003 0.04 0.02 0.029 0.04 0.029 0.039 0.05 0.06 0.02 0.003 0.04 0.029 0.039 0.05 0.06 0.02 0.003 0.04 0.02 0.029 0.039 0.05 0.06 0.003 0.04 0.02 0.029 0.039 0.05 0.06 0.003

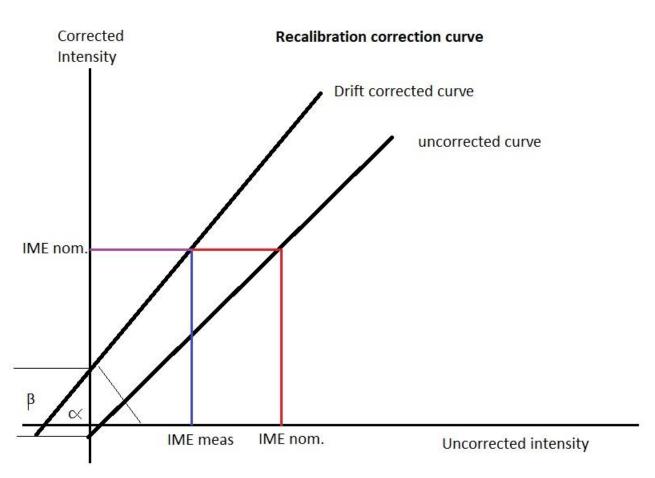
result comparison to previously
measured set of calibration samples
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Emission spectrometry is a relative measurement method





Recalibration, Standardization or Drift correction









Type Calibration

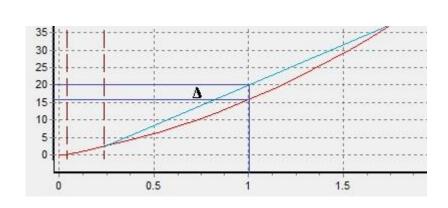


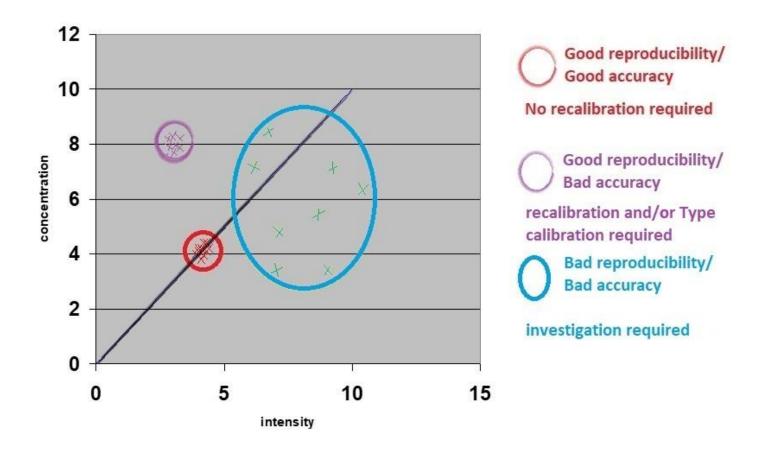
Illustration of exaggerated concentration error Δ in a calibration curve where Type calibration correction can be used to improve accuracy.







Measurement accuracy and reproducibility







The analytical Process











Sample taking





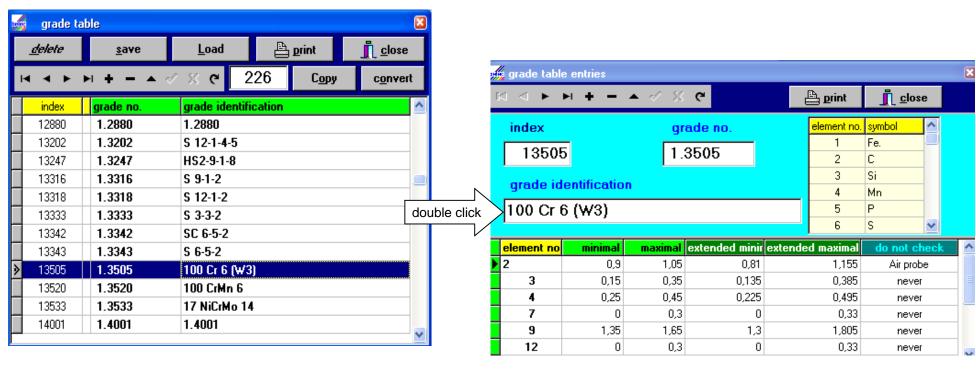
Type Calibration and measure program







Material Data Base / Grade Table



- software compares analysis results with element concentration ranges and reports grade maches
- virtually unlimited material data entries
- international material standards as well as customised data base can be loaded
- Specific material data base can be linked to a measuring program





Thank you.

For more information or feedback contact Glen Thiele

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- References
- (1) V. Thomsen, Modern Spectrochemical Analysis of Metals: An Introduction for Users of Arc/Spark Instrumentation (ASM International, Materials Park, Ohio, 1996).
- (2) Power Point Presentations definitions and images: Belec Spectrometrie Opto-Electronik (2012 2016).
- (3) Various definitions and images: Wikipedia.
- (4) Melting Spectro Analysis Version 6: G Henderieckx Gietech BV